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Before you answer any further  
questions, please turn on the heat and

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watch the molecules and the graph.  
Think about the definitions of the  
different phases of matter (solid, ...  
PHASE CHANGE: Simulation worksheet 2.  
What was difficult about the procedure  
(measuring the temperature, keeping  
the time, ...

## **PHASE CHANGE: Simulation**

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## **worksheet**

Phase Change Worksheet The graph was drawn from data collected as a substance was heated at a constant rate. Use the graph to answer the following questions. 1 80 Phæe Change 60 1 40 1 20 100 60 40 12 Tim e (minutes) 20 22 At point A, the beginning of observations, the

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substance exists in a solid state.  
Material in this phase has volume and  
shape.

## **Nicolet High School**

Heat with Phase Change Worksheet 1)  
How many joules are required to heat  
250 grams of liquid water from 00to  
1000C ? 2) How many joules are

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required to melt 100 grams of water? 3)  
How many joules are required to boil  
150 grams of water?

## **Heat with Phase Change Worksheet**

The phase change from water vapor to liquid water is known as... Preview this quiz on Quizizz. The phase change from water vapor to liquid water is known

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as... Phase Changes DRAFT. ... answer  
choices . evaporation. precipitation.  
condensation. sublimation. Tags:  
Question 2 . SURVEY . 60 seconds . Q.

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KEY PHASE DIAGRAM WORKSHEET Part A

- Generic Phase Diagram. Answer the questions below in relation to the following generic phase diagram. 1.

Which section represents the solid phase? A 2. What section represents the liquid phase? C 3. What section represents the gas phase? B 4.

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Calculating Energy for Changes of Phase  
To Calculate Heat: Heat to Change the  
Temperature:  $Q = m \cdot T \cdot s$  Heat to  
Melt:  $Q = m \cdot H_{\text{fus}}$  Heat to Boil:  $Q = m$   
 $\cdot H_{\text{vap}}$  For Water:  $H_{\text{fus}} = 334 \text{ J/g}$   $H_{\text{vap}} = 2260 \text{ J/g}$   
Specific Heat: ice =  $2.06 \text{ J/g}^\circ\text{C}$   
water =  $4.184 \text{ J/g}^\circ\text{C}$  steam =  $2.03 \text{ J/g}^\circ\text{C}$

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## **Calculating Energy for Changes of Phase**

(answer in terms of potential and/or kinetic energy) \_\_\_\_\_ The energy is causing the position of the atoms to change (P.E. is increasing) there is a phase change \_\_\_\_\_ c) What phase change, if any, is taking place?

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## **Heating Curve Worksheet - Studylib**

Question: Worksheet Directios Fer H  
Hange Only, With No Phase Change  
Then Name: Per: USE Q-m CAT Change  
Involved, Temp. Remains Constant Then  
USE Q-m AHas For Melting (E Absorbed)  
Or Treezing CE . If Phase Involved,  
Released) If Temp, Crosses F Or USE Q-

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m AHap For Boiling (E Absorbed) Or  
Condensing (E Released) Ranges  
Related Hom Gas To Liquid To Solid  
Then Must ...

## **Worksheet Directios Fer H Hange Only, With No Phas ...**

A student worksheet (Phase Change  
Extend) can be found on Page 8. Note: If

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students have not yet earned Achievements, remind them that they need to select the Achievement first, in order to receive credit for completion.

### **Collisions Lesson Plan Energy & Phase Change**

Changes of State Worksheet Write the correct phase change based on the

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diagram at the top of the page.  
(snowman, pot of boiling water, floating  
in the water) 1. Sublimation 2. Melting 3.  
Evaporation 4. Solidification 5.  
Condensation 6. Deposition 7. Complete  
the table, using the examples inserted  
below. Change of state Heat added or  
released?



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